

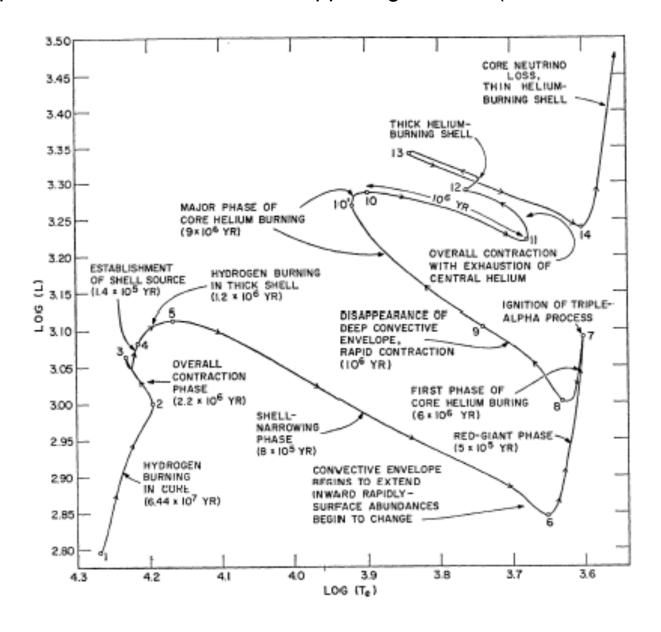
As hydrogen is exhausted in the (convective) core of a star (point 2) it moves away from the main sequence (point 3)

What happens to the star?

- lower T → redder
- same L → larger (Stefan's L.)

star becomes a **red giant** 

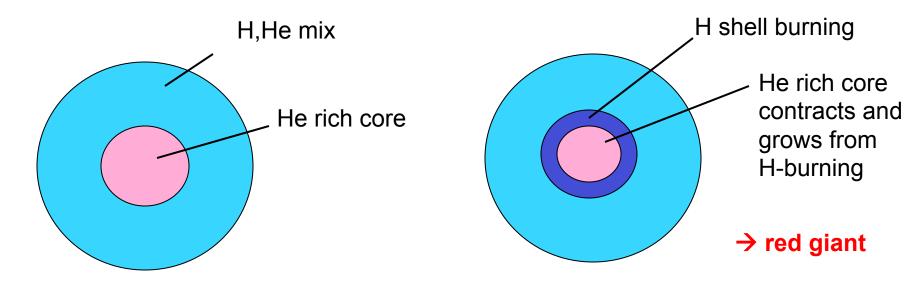
For completeness – here's what's happening in detail (5 solar mass ZAMS star):



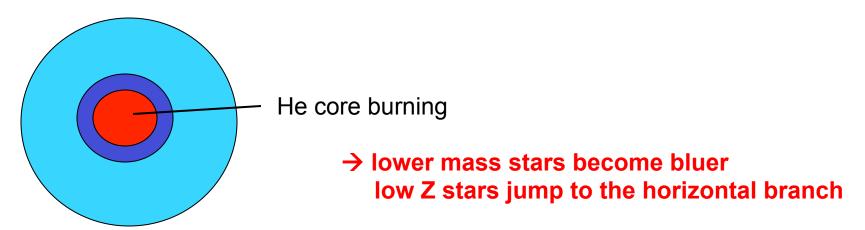
### What happens at hydrogen exhaustion

(assume star had convective core)

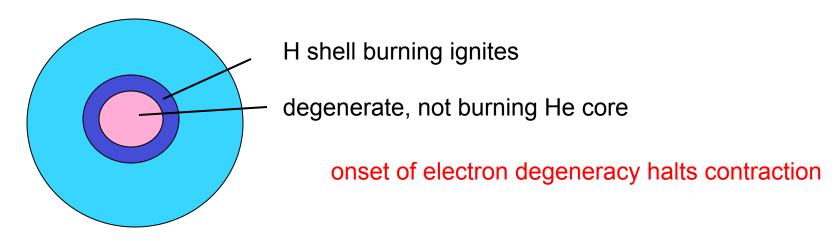
### 1. Core contracts and heats



### 2. Core He burning sets in



### 2. a (M < 2.25 $M_0$ ) Degenerate He core



then He core grows by H-shell burning until He-burning sets in.

→ He burning is initially unstable (He flash)

in degenerate electron gas, pressure does not depend on temperature (why?) therefore a slight rise in temperature is not compensated by expansion

→ thermonuclear runaway:

- rise temperature
- accelerate nuclear reactions
- increase energy production

## Why does the star expand and become a red giant?

Because of higher Coulomb barrier He burning requires much higher temperatures

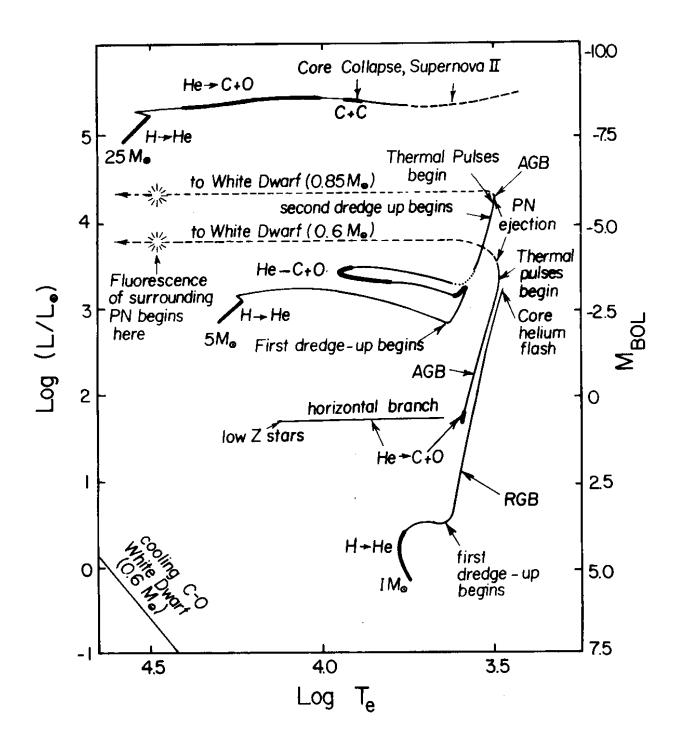
- → drastic change in central temperature
- → star has to readjust to a new configuration

### Qualitative argument:

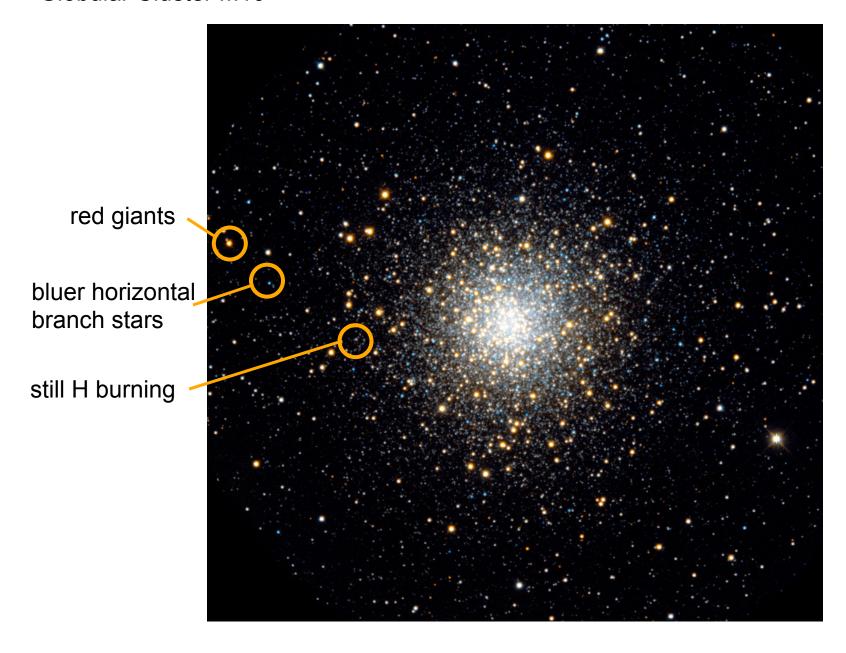
- need about the same Luminosity similar temperature gradient dT/dr
- now much higher T<sub>c</sub> need larger star for same dT/dr

Lower mass stars become red giants during shell H-burning

If the sun becomes a red giant in about 5 Bio years, it will almost fill the orbit of Mars



### Globular Cluster M10



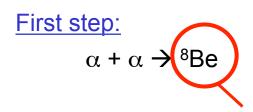
# He burning overview

- Lasts about 10% of H-burning phase
- Temperatures: ~300 Mio K
- Densities ~ 10<sup>4</sup> g/cm<sup>3</sup>

Reactions: 
$${}^{4}\text{He} + {}^{4}\text{He} + {}^{4}\text{He} \rightarrow {}^{12}\text{C}$$
 (triple  $\alpha$  process)  ${}^{12}\text{C} + {}^{4}\text{He} \rightarrow {}^{16}\text{O}$  ( ${}^{12}\text{C}(\alpha,\gamma)$ )

Main products: carbon and oxygen (main source of these elements in the universe)

# Helium burning 1 – the $3\alpha$ process



unbound by ~92 keV – decays back to 2  $\alpha$  within 2.6E-16 s !

but small equilibrium abundance is established

### Second step:

<sup>8</sup>Be +  $\alpha \rightarrow$  <sup>12</sup>C\* would create <sup>12</sup>C at excitation energy of ~7.7 MeV

1954 Fred Hoyle (now Sir Fred Hoyle) realized that the fact that there is carbon in the universe requires a resonance in <sup>12</sup>C at ~7.7 MeV excitation energy

1957 Cook, Fowler, Lauritsen and Lauritsen at Kellogg Radiation Laboratory at Caltech discovered a state with the correct properties (at 7.654 MeV)

Experimental Nuclear Astrophysics was born

# How did they do the experiment?

- Used a deuterium beam on a 11B target to produce 12B via a (d,p) reaction.
- 12B  $\beta$ -decays within 20 ms into the second excited state in 12C
- This state then immediately decays under alpha emission into 8Be
- Which immediately decays into 2 alpha particles

So they saw after the delay of the b-decay 3 alpha particles coming from their target after a few ms of irradiation

This proved that the state can also be formed by the 3 alpha process ...

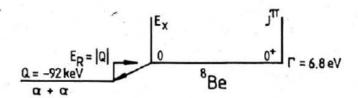
- → removed the major roadblock for the theory that elements are made in stars
- → Nobel Prize in Physics 1983 for Willy Fowler (alone!)



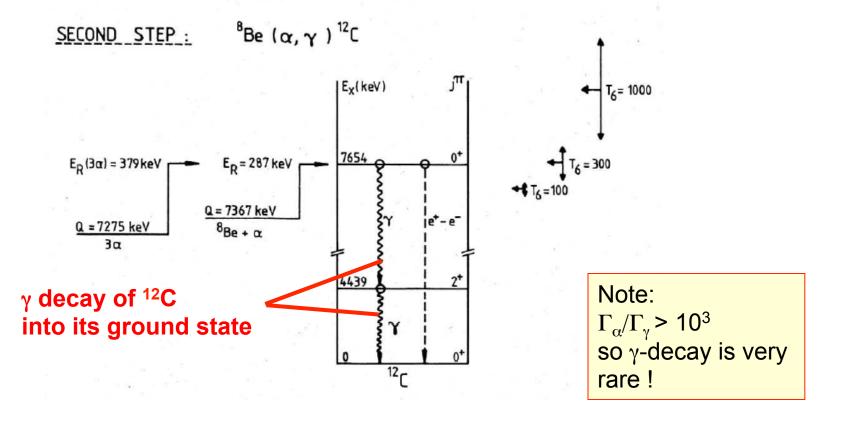


### Third step completes the reaction:

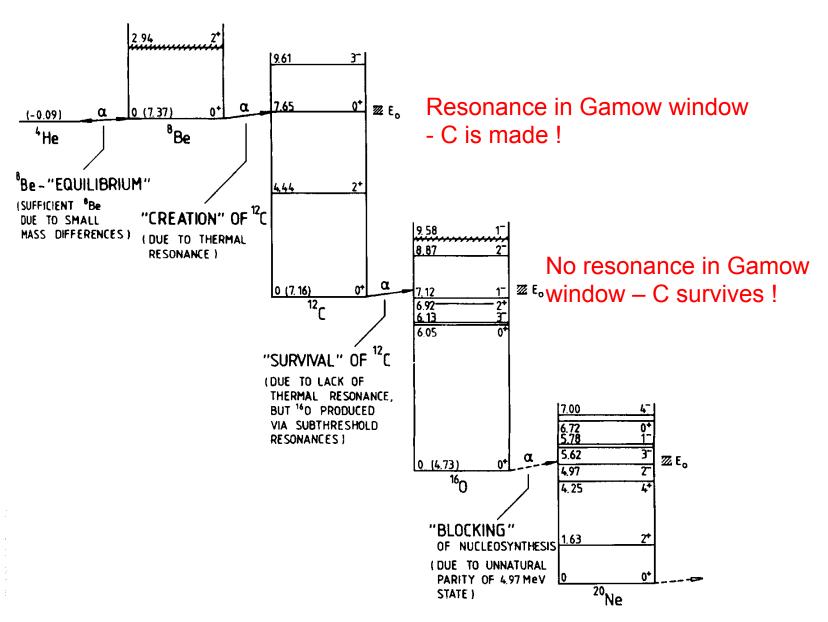
FIRST STEP:  $\alpha + \alpha = {}^{8}Be$ 

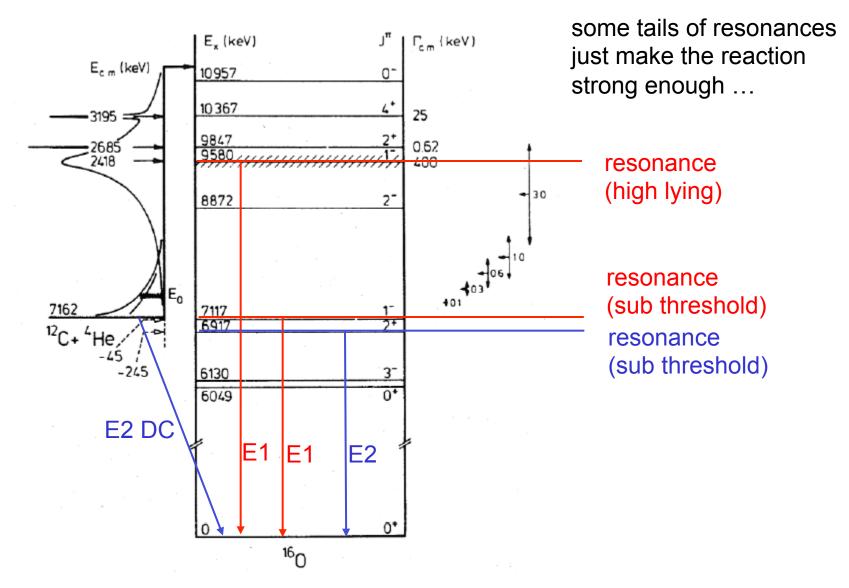


Note:  $^8$ Be ground state is a 92 keV resonance for the  $\alpha$ + $\alpha$  reaction



# Helium burning 2 – the $^{12}C(\alpha,\gamma)$ rate





complications:

- very low cross section makes direct measurement impossible
- subthreshold resonances cannot be measured at resonance energy
- Interference between the E1 and the E2 components

#### Therefore:

Uncertainty in the  $^{12}$ C( $\alpha,\gamma$ ) rate is the single most important nuclear physics uncertainty in astrophysics

#### Affects:

- C/O ration → further stellar evolution (C-burning or O-burning?)
- iron (and other) core sizes (outcome of SN explosion)
- Nucleosynthesis (see next slide)

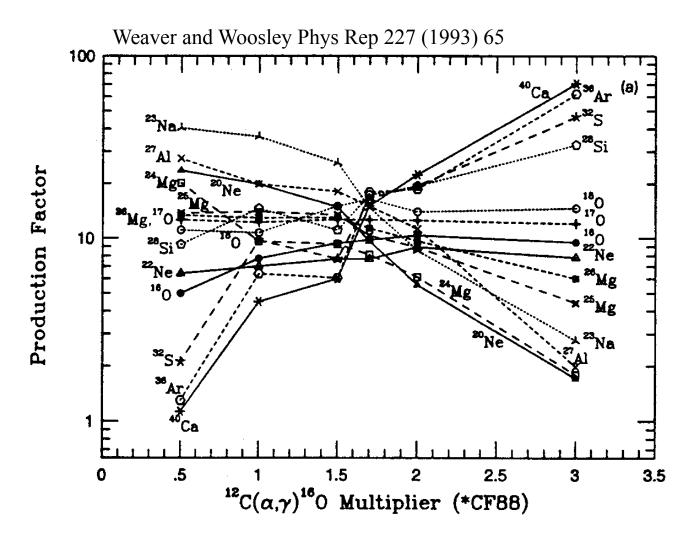
### Some current results for S(300 keV):

S<sub>E2</sub>=53+13-18 keV b (Tischhauser et al. PRL88(2002)2501

S<sub>E1</sub>=79+21-21 keV b (Azuma et al. PRC50 (1994) 1194)

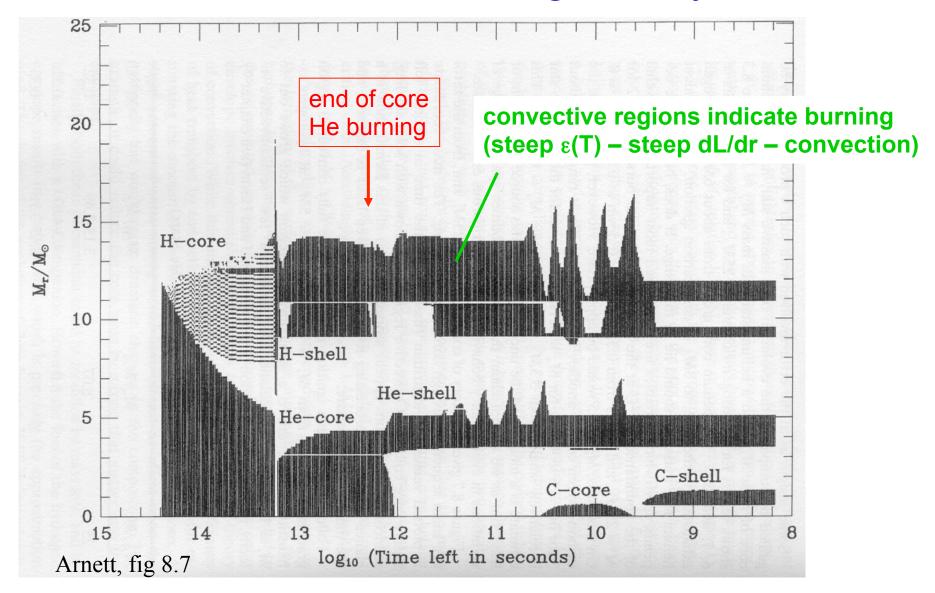
But others range among groups larger!

### Massive star nucleosynthesis model as a function of ${}^{12}C(\alpha,\gamma)$ rate

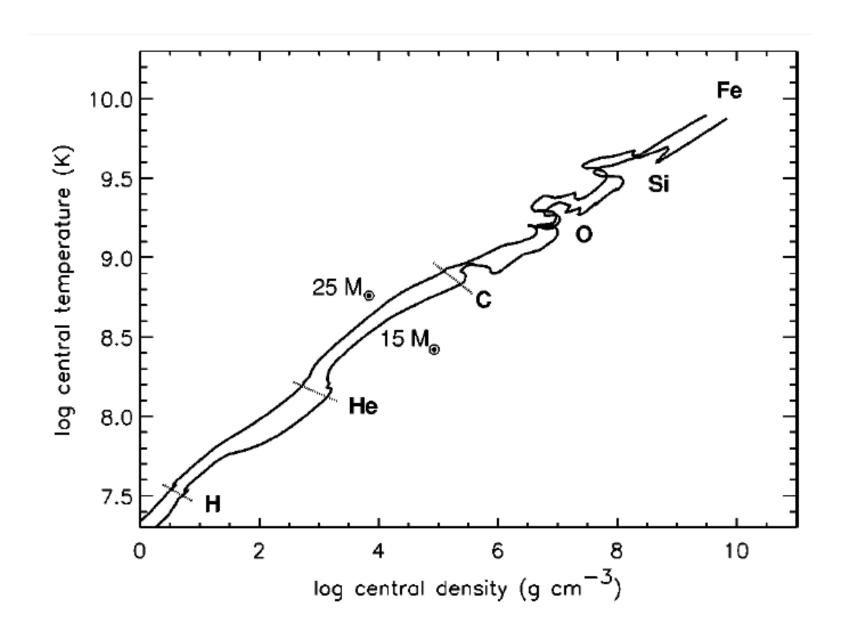


- This demonstrates the sensitivity
- One could deduce a preference for a total S(300) of ~120-220 (But of course we cannot be sure that the astrophysical model is right)

# End of core helium burning and beyond



→ note complicated multiple burning layers !!!



# Carbon burning

### **Burning conditions:**

### for stars > 12 M<sub>o</sub> (solar masses) (ZAMS)

T~ 0.6-1.2 Bio K  $\rho$  ~ 3x10<sup>5</sup> g/cm<sup>3</sup>

Reactions: 
$${}^{12}C + {}^{12}C \rightarrow {}^{23}Mg + n - 2.62 \text{ MeV (<1\%)}$$

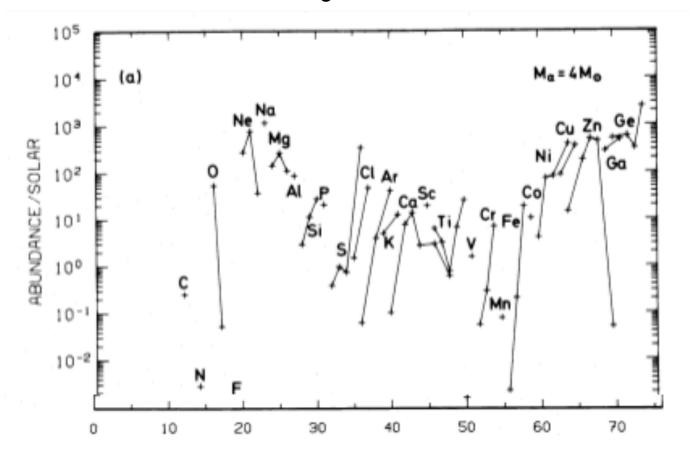
$$^{20}$$
Ne +  $\alpha$  + 4.62 MeV (50%)

$$^{23}$$
Na + p + 2.24 MeV (50%)

Plus 100s of other  $p,\alpha,n$  induced reactions Including a mild s-process

Main products of carbon burning: <sup>16</sup>O (a survivor from helium burning), <sup>20,21,22</sup>Ne, <sup>23</sup>Na, <sup>24,25,26</sup>Mg, and <sup>26,27</sup>Al

### Composition after core carbon burning:



# Neon burning

### **Burning conditions:**

### for stars > 12 M<sub>o</sub> (solar masses) (ZAMS)

T~ 1.3-1.7 Bio K 
$$\rho$$
 ~ 10<sup>6</sup> g/cm<sup>3</sup>

### Why would neon burn before oxygen ???

#### Answer:

Temperatures are sufficiently high to initiate **photodisintegration** of <sup>20</sup>Ne

$$\left.\begin{array}{c} ^{20}\text{Ne+}\gamma \rightarrow ^{16}\text{O} + \alpha \\ ^{16}\text{O} + \alpha \rightarrow ^{20}\text{Ne} + \gamma \end{array}\right\} \quad \text{equilibrium is established}$$

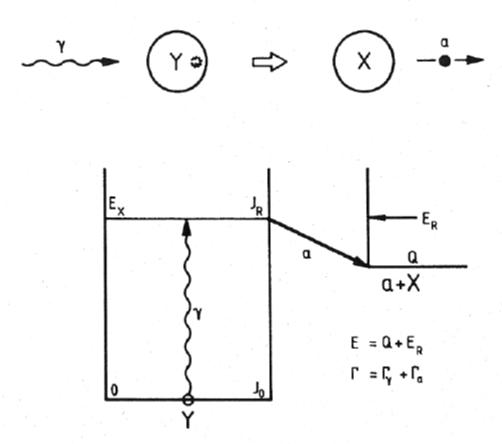
this is followed by (using the liberated helium)

$$^{20}$$
Ne+ $\alpha \rightarrow ^{24}$ Mg +  $\gamma$ 

so net effect: 
$$2^{20}\text{Ne} \rightarrow {}^{16}\text{O} + {}^{24}\text{Mg} + 4.59 \text{ MeV}.$$

# Photodisintegration

## PHOTODISINTEGRATION $Y(\gamma,\alpha)X$



## Calculations of inverse reaction rates

A reaction rate for a process like  $^{20}\text{Ne+}\gamma \rightarrow ^{16}\text{O} + \alpha$  can be easily calculated from the inverse reaction rate  $^{16}\text{O}+\alpha \rightarrow ^{20}\text{Ne} + \gamma$  using the formalism developed so far.

In general there is a simple relationship between the rates of a reaction rate and its inverse process (if all particles are thermalized)

### **Derivation of "detailed balance principle":**

Consider the reaction A+B → C with Q-value Q in thermal equilibrium. Then the abundance ratios are given by the Saha equation:

$$\frac{n_{A}n_{B}}{n_{C}} = \frac{g_{A}g_{B}}{g_{C}} \left(\frac{m_{A}m_{B}}{m_{C}}\right)^{3/2} \left(\frac{kT}{2\pi\hbar^{2}}\right)^{3/2} e^{-Q/kT}$$

In equilibrium the abundances are constant per definition. Therefore in addition

$$\frac{dn_C}{dt} = n_A n_B < \sigma v > -\lambda_C n_C = 0$$

or 
$$\frac{\lambda_C}{<\sigma \upsilon>} = \frac{n_A n_B}{n_C}$$

If  $\langle \sigma v \rangle$  is the A+B  $\rightarrow$  C reaction rate, and  $\lambda_C$  is the C  $\rightarrow$ A+B decay rate Therefore the rate ratio is defined by the Saha equation as well! Using both results one finds

$$\frac{\lambda_C}{\langle ov \rangle} = \frac{g_A g_B}{g_C} \left(\frac{m_A m_B}{m_C}\right)^{3/2} \left(\frac{kT}{2\pi\hbar^2}\right)^{3/2} e^{-Q/kT}$$

or using  $m_C \sim m_A + m_B$  and introducing the reduced mass  $\mu$ 

#### **Detailed balance:**

$$\frac{\lambda_C}{\langle \sigma v \rangle} = \frac{g_A g_B}{g_C} \left(\frac{\mu kT}{2\pi\hbar^2}\right)^{3/2} e^{-Q/kT}$$

So just by knowing partition functions g and mass m of all participating particles on can calculate for every reaction the rate for the inverse process.

#### **Partition functions:**

For a particle in a given state i this is just  $g_i = 2J_i + 1$ 

However, in an astrophysical environment some fraction of the particles can be in thermally excited states with different spins. The partition function is then given by:

$$g = \sum_{i} g_{i} e^{-E_{i}/kT}$$

# Silicon burning

### **Burning conditions:**

```
T~ 3-4 Bio \rho \sim 10^9 \text{ g/cm}^3
```

### **Reaction sequences:**

- Silicon burning is fundamentally different to all other burning stages.
- Complex network of fast  $(\gamma,n)$ ,  $(\gamma,p)$ ,  $(\gamma,a)$ ,  $(n,\gamma)$ ,  $(p,\gamma)$ , and  $(a,\gamma)$  reactions
- The net effect of Si burning is: 2 <sup>28</sup>Si --> <sup>56</sup>Ni,

### need new concept to describe burning:

**Nuclear Statistical Equilibrium (NSE)** 

**Quasi Statistical Equilibrium (QSE)** 

# Nuclear Statistical Equilibrium

#### **Definition:**

In NSE, each nucleus is in equilibrium with protons and neutrons

Means: the reaction Z \* p + N \* n <---> (Z,N) is in equilibrium

Or more precisely: 
$$Z \cdot \mu_p + N \cdot \mu_n = \mu_{(Z,N)}$$
 for all nuclei (Z,N)

NSE is established when both, photodisintegration rates of the type

$$(Z,N) + \gamma --> (Z-1,N) + p$$

$$(Z,N) + \gamma --> (Z,N-1) + n$$

$$(Z,N) + \gamma$$
 -->  $(N-2,N-2) + \alpha$ 

and capture reactions of the types

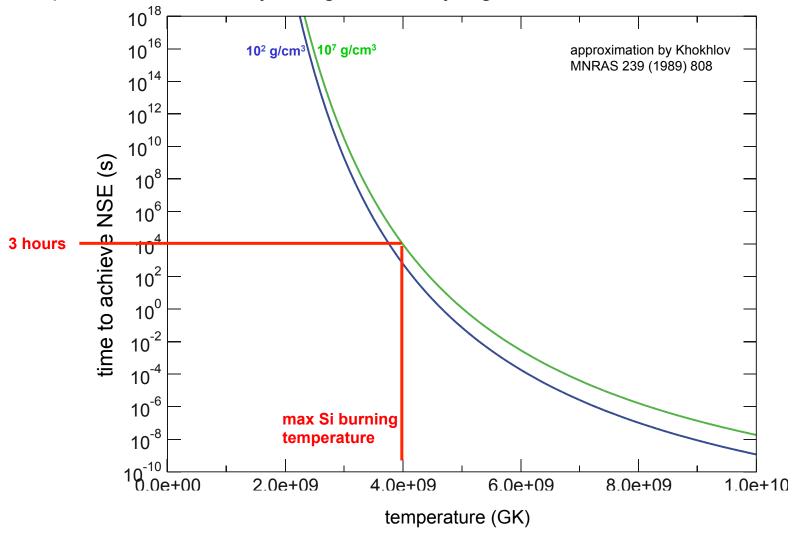
$$(Z,N) + p --> (Z+1,N)$$

$$(Z,N) + n \longrightarrow (Z,N+1)$$

$$(Z,N) + \alpha \longrightarrow (Z+2,N+2)$$

NSE is established on the timescale of these reaction rates (the slowest reaction)

A system will be in NSE if this timescale is shorter than the timescale for the temperature and density being sufficiently high.



### **Nuclear Abundances in NSE**

The ratio of the nuclear abundances in NSE to the abundance of free protons and neutrons is entirely determined by

$$Z \cdot \mu_p + N \cdot \mu_n = \mu_{(Z,N)}$$

which only depends on the chemical potentials

$$\mu = mc^2 + kT \ln \left[ \frac{n}{g} \left( \frac{h^2}{2\pi mkT} \right)^{3/2} \right]$$

So all one needs are density, temperature, and for each nucleus mass and partition function (one does not need reaction rates !! - except for determining whether equilibrium is indeed established)

Solving the two equations on the previous page yields for the abundance ratio:

$$Y(Z,N) = Y_p^Z Y_n^N G(Z,N) (\rho N_A)^{A-1} \frac{A^{3/2}}{2^A} \left( \frac{2\pi \hbar^2}{m_u kT} \right)^{\frac{3}{2}(A-1)} e^{B(Z,N)/kT}$$

with the nuclear binding energy B(Z,N)

### **Some features of this equation:**

- in NSE there is a mix of free nucleons and nuclei
- higher density favors (heavier) nuclei
- higher temperature favors free nucleons (or lighter nuclei)
- nuclei with high binding energy are strongly favored

To solve for Y(Z,N) two additional constraints need to be taken into account:

Mass conservation 
$$\sum_{i} A_{i}Y_{i} = 1$$

Proton/Neutron Ratio 
$$\sum_{i} Z_{i} Y_{i} = Y_{e}$$

In general, weak interactions are much slower than strong interactions. Changes in  $\rm Y_e$  can therefore be calculated from beta decays and electron captures on the NSE abundances for the current, given  $\rm Y_e$ 

In many cases weak interactions are so slow that Y<sub>e i</sub>is roughly fixed.

## Sidebar – another view on NSE: Entropy

### In Equilibrium the entropy has a maximum dS=0

• This is equivalent to our previous definition of equilibrium using chemical potentials: First law of thermodynamics:

$$dE = TdS + \rho N_A \sum_{i} \mu_i dY_i - pdV$$

so as long as dE=dV=0, we have in equilibrium (dS=0):

$$\sum_{i} \mu_{i} dY_{i} = 0$$

for any reaction changing abundances by dY

For Zp+Nn --> (Z,N) this yields again

$$Z \cdot \mu_p + N \cdot \mu_n = \mu_{(Z,N)}$$

There are two ways for a system of nuclei to increase entropy:

- 1. Generate energy (more Photon states) by creating heavier, more bound nuclei
- 2. Increase number of free nucleons by destroying heavier nuclei

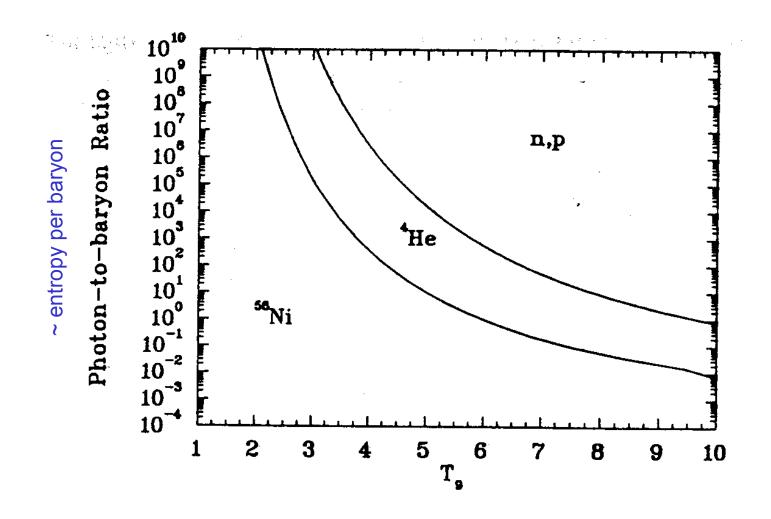
These are conflicting goals, one creating heavier nuclei around iron/nickel and the other one destroying them

- The system settles in a compromise with a mix of nucleons and most bound nuclei
- Tendency:

```
high entropy per baryon (low \rho, high T) \rightarrow more nucleons low entropy per baryon (high \rho, low T) \rightarrow more heavy nuclei
```

(entropy per baryon (if photons dominate):  $\sim T^3/\rho$ )

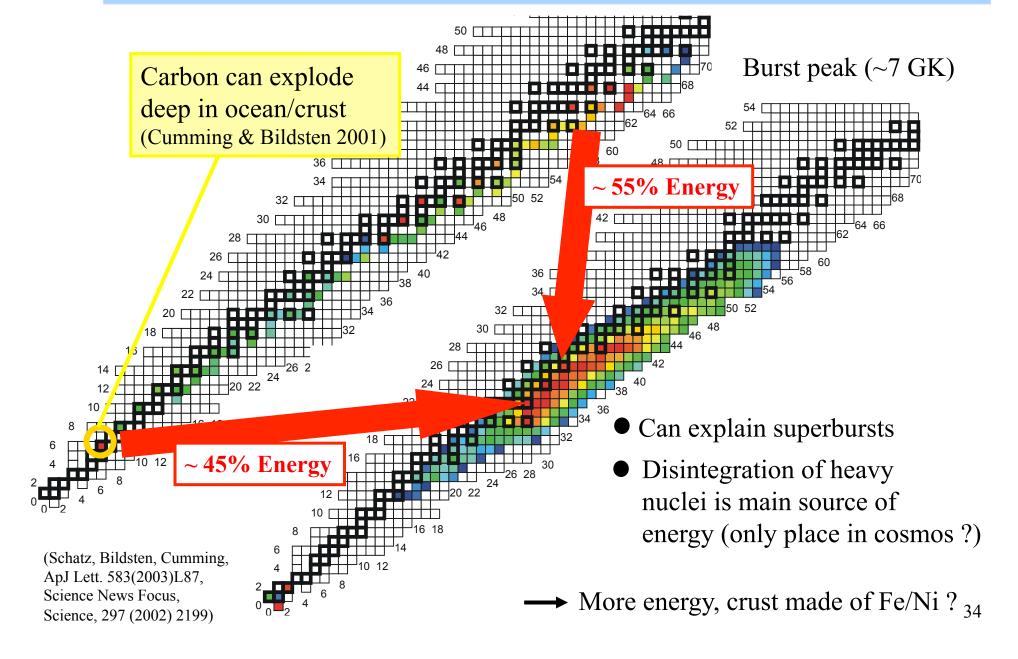
### NSE composition (Y<sub>e</sub>=0.5)







## Nuclear physics during superburst



## Incomplete Equilibrium - Equilibrium Cluster

Often, some, but not all nuclei are in equilibrium with protons and neutrons (and with each other).

A group of nuclei in equilibrium is called an equilibrium cluster. Because of reactions involving single nucleons or alpha particles being the mediators of the equilibrium, neighboring nuclei tend to form equilibrium clusters, with cluster boundaries being at locations of exceptionally slow reactions.

This is referred as Quasi Statistical Equilibrium (or QSE)

### **Typical Example:**

 $3\alpha$  rate is slow  $\longrightarrow$   $\alpha$  particles are not in full NSE

## **NSE** during Silicon burning

- Nuclei heavier than <sup>24</sup>Mg are in NSE
- High density environment favors heavy nuclei over free nucleons
- Y<sub>e</sub> ~0.46 in core Si burning due to some electron captures
  - **→** main product <sup>56</sup>Fe (26/56 ~ 0.46)

#### formation of an iron core

(in explosive Si burning no time for weak interactions,  $Y_e \sim 0.5$  and therefore final product <sup>56</sup>Ni)

# Summary stellar burning

TABLE 8.1 Evolutionary Stages of a 25 M . Stara

	Stage	Time Scale	Temperature $(T_9)$	Density (g cm <sup>-3</sup> )
>0.8M <sub>0</sub> ↓ ↓	Hydrogen burning	$7 \times 10^{6} \text{ y}$	0.06	5
	Helium burning	$5 \times 10^5 \text{ y}$	0.23	$7 \times 10^2$
	Carbon burning	600 y	0.93	$2 \times 10^{5}$
	Neon burning	1 y	1.7	$4 \times 10^{6}$
	Oxygen burning	6 months	2.3	$1 \times 10^{7}$
	Silicon burning	1 d	4.1	$3 \times 10^{7}$
	Core collapse	seconds	8.1	$3 \times 10^{9}$
	Core bounce	milliseconds	34.8	$\simeq 3 \times 10^{14}$
	Explosive burning	0.1-10 s	1.2-7.0	Varies

Why do timescales get smaller?

**Note:** Kelvin-Helmholtz timescale for red supergiant ~10,000 years, so for massive stars, no surface temperature - luminosity change for C-burning and beyond

### Final composition of a 25 $M_0$ star:

